

Available online at www.sciencedirect.com



Geochimica et Cosmochimica Acta

Geochimica et Cosmochimica Acta 71 (2007) 5033–5043

www.elsevier.com/locate/gca

# Seasonal and diurnal variations of  ${}^{13}C^{18}O^{16}O$  in air: Initial observations from Pasadena, CA

Hagit P. Affek <sup>a,\*</sup>, Xiaomei Xu <sup>b</sup>, John M. Eiler <sup>a</sup>

a Division of Geological and Planetary Sciences, California Institute of Technology, MC100-23, Pasadena, CA 91125, USA b Department of Earth System Sciences, University of California, Irvine, CA 92697-3100, USA

Received 26 March 2007; accepted in revised form 15 August 2007; available online 14 September 2007

## Abstract

The budget of atmospheric CO<sub>2</sub> is widely studied using records of temporal and spatial variations of concentrations,  $\delta^{13}$ C and  $\delta^{18}$ O values. However, the number and diversity of sources and sinks prevents these alone from fully constraining the budget. Molecules containing two rare isotopes can serve as additional tracers and potentially provide additional, independent, constraints. We present data documenting seasonal and diurnal variations of  $CO_2$  having a mass of 47 u (mostly  $^{13}C^{18}O^{16}O$ ) in air from Pasadena, CA. We report these data using the 'mass 47 anomaly' ( $\Delta_{47}$ ation of  $R^{47}$ (=[47]/[44]) from that expected for a random distribution of isotopologues. Between February 2004 and December 2005,  $\Delta_{47}$  showed a seasonal pattern that differed significantly from that expected based on thermodynamic equilibrium. During the year 2004  $\Delta_{47}$  was 0.76‰ in winter, increased to 0.87‰ in summer and gradually decreased through the autumn to 0.81‰ at the end of the year.  $\Delta_{47}$  then increased again through the winter and spring of 2005 to 0.97‰ in summer followed by a decrease to 0.88‰ at the end of 2005. The seasonal variations cannot be accounted for by variations in the relative contribution of local fossil fuel sources. Diurnal variations were the combined effect of both fuel combustion and respiration having  $\Delta_{47}$  values of 0.41‰ and ca. 0.77‰, respectively. The seasonal cycle may be interpreted as a competition between low  $\Delta_{47}$ values in respiration and higher  $\Delta_{47}$  values resulting from CO<sub>2</sub>-water exchange in photosynthesis.  $© 2007 Elsevier Ltd. All rights reserved.$ 

# 1. INTRODUCTION

The budget of atmospheric  $CO<sub>2</sub>$  is widely studied using temporal and spatial records of concentration and  $\delta^{13}$ C and  $\delta^{18}$ O values combined with estimates of the values of these isotopic tracers associated with  $CO<sub>2</sub>$  fluxes [\(Francey](#page-10-0) [and Tans, 1987; Francey et al., 1995; Ciais et al., 1995a,b,](#page-10-0) [1997; Peylin et al., 1999; Cuntz et al., 2003\)](#page-10-0). In particular,  $\delta^{13}$ C values are primarily used to partition marine from ter-restrial fluxes (e.g., [Ciais et al., 1995a\)](#page-10-0) and  $\delta^{18}$ O values are primarily used to separate terrestrial net photosynthesis from soil respiration ([Yakir and Wang, 1996; Ciais et al.,](#page-10-0) [1997; Peylin et al., 1999\)](#page-10-0). Latitudinal gradients, long-term

change and seasonal variations are commonly used in isotopic mass balance models to partition fluxes. However, the diversity and variability of fluxes to and from the atmosphere prevents these alone from fully constraining the budget. Molecules containing two rare isotopes potentially provide additional, independent constraints. We present data documenting seasonal and diurnal variations of mass of 47 u (mostly  $13C^{18}O^{16}O$ ) in air from Pasadena, CA. We report these data using the 'mass 47 anomaly'  $(\Delta_{47})$ , which reflects the deviation of the abundance of mass 47 isotopologues from that predicted for a random distribution of isotopes and is a measure of the preference of two heavy isotopes  $(^{13}C$  and  $^{18}O$ ) to 'clump' together in one chemical bond [\(Eiler and Schauble, 2004; Wang et al.,](#page-10-0) [2004; Affek and Eiler, 2006](#page-10-0)).

Theoretical calculation (verified experimentally for the temperature range of 267–303 K; [Eiler et al., 2003\)](#page-10-0) suggest that at thermodynamic equilibrium the  $\Delta_{47}$  value of CO<sub>2</sub> is temperature dependent ([Eiler and Schauble, 2004; Wang](#page-10-0)

Corresponding author. Present address: Department of Geology and Geophysics, Yale University, P.O. Box 208109, New Haven, CT 06520-8109, USA.

E-mail addresses: [hagit.affek@yale.edu](mailto:hagit.affek@yale.edu) (H.P. Affek), [xxu@uci.](mailto:xxu@uci. edu) [edu](mailto:xxu@uci. edu) (X. Xu), [eiler@gps.caltech.edu](mailto:eiler@gps.caltech.edu) (J.M. Eiler).

<sup>0016-7037/\$ -</sup> see front matter © 2007 Elsevier Ltd. All rights reserved. doi:10.1016/j.gca.2007.08.014

[et al., 2004\)](#page-10-0).  $CO_2-H_2O$  isotope exchange drives  $CO_2$  toward thermodynamic equilibrium and is likely to result in  $\Delta_{47}$ values that reflects the temperature of exchange, with  $\Delta_{47}$ of 0.95‰ at 25 °C [\(Eiler and Schauble, 2004](#page-10-0)). Fluxes that involve  $CO_2$  equilibrium with water and whose  $\delta^{18}O$  values reflects these equilibria, such as exchange with ocean or leaf water, although never directly characterized for  $\Delta_{47}$ , are expected to have  $\Delta_{47}$  values close to the thermodynamic equilibrium values. However,  $\Delta_{47}$  in human breath was observed to be 0.76‰ [\(Affek and Eiler, 2006](#page-10-0)), lower than the expected at thermodynamic equilibrium at body temperature (0.86‰ at 37 °C).  $CO<sub>2</sub>$  in car exhaust was found to have a  $\Delta_{47}$  value of 0.41‰, presumably reflecting higher temperature equilibrium quenched in the exhaust stream ([Affek and Eiler, 2006](#page-10-0)).

 $\Delta_{47}$  values in ambient air must reflect a balance of CO<sub>2</sub> fluxes and their corresponding  $\Delta_{47}$  values, including both combustion sources and sources that involve exchange between the atmosphere and either ocean water or the land biosphere. The work presented here examined the stable isotope composition of urban air in Pasadena, CA, aimed at exploring the systematics and causes of temporal variation in  $\Delta_{47}$  values. The typical synoptic air pattern in the Los Angeles basin is a combination of sea-breeze and mountain flow that leads to air flow from the coast inland, pushing polluted urban air towards the San Gabriel mountains [\(Lu and Turco, 1995](#page-10-0)). Pasadena is located at the foothills of these mountains. It is therefore highly affected by the plume of polluted air, and the  $CO<sub>2</sub>$  in Pasadena air is likely to contain a significant component originating from combustion sources.

#### 2. METHODS

### 2.1. Air sampling and isotopic analysis

We sampled air (a sample size of 5 L; air was dried by  $Mg(CIO<sub>4</sub>)<sub>2</sub>$  during sampling) in Pasadena, CA, between February 2004 and December 2005. Samples were generally obtained at  $\sim$ 10am from a third floor balcony at the Caltech campus (34.137 $\textdegree$ N, 118.128 $\textdegree$ W), facing a courtyard in which the dominant vegetation is deciduous Sweetgum trees and lawn grass. The sampling time of day is shortly after the decrease in  $CO<sub>2</sub>$  concentration following the peak associated with morning traffic when the concentrations are already close to the diurnal minimum, suggesting minimal contribution from local sources. However, the air samples do not reflect background air (that is, air far from urban centers or vigorously active terrestrial ecosystems), because regional contributions from anthropogenic fluxes in the greater Los Angeles basin cannot be avoided at any time of the day. Diurnal variations were examined by collecting air every 3 h over a 24 h cycle in February 14, February 21 and March 24, 2006. In order to estimate the  $\Delta_{47}$  values in background air, samples from Cape Grim, Tasmania (40.68°S, 144.68°E) and Barrow, AK (71.32°N, 156.6°W) were analyzed. Cape Grim air was sampled between 1 and 4pm, when wind direction was west to south-west. Air in Barrow was sampled between 8am and 2pm, when the wind direction was west to south-west.  $CO<sub>2</sub>$  was extracted from the collected air by freezing in liquid  $N_2$ on a glass vacuum line [\(Affek and Eiler, 2006\)](#page-10-0).

The isotopic composition of the extracted  $CO<sub>2</sub>$  was measured by a dual inlet gas source mass spectrometry using a Finnigan-MAT 253 isotope ratio mass spectrometer, configured to measure masses 44, 45, 46, 47, 48 and 49 to obtain  $\delta^{13}C$ ,  $\delta^{18}O$  and  $\Delta_{47}$  values.  $\Delta_{47}$  is calculated as:

$$
\Delta_{47} = \left[ \frac{R^{47}}{2R^{13} \cdot R^{18} + 2R^{17} \cdot R^{18} + R^{13} \cdot (R^{17})^2} - \frac{R^{46}}{2R^{18} + 2R^{13} \cdot R^{17} + (R^{17})^2} - \frac{R^{45}}{R^{13} + 2R^{17}} + 1 \right] \cdot 1000
$$
\n(1)

and the  $\Delta_{47}$  calibration is obtained by measuring CO<sub>2</sub> heated to  $1000 \, \text{°C}$ , which is assumed to have a stochastic distribution of isotopes among all isotopologues ([Eiler](#page-10-0) [and Schauble, 2004;](#page-10-0) a detailed description of the method is given in [Affek and Eiler, 2006](#page-10-0)). Some of the air samples (during October 2004–December 2005) were analyzed also for  $^{14}$ C content at the University of California, Irvine (using the method described in [Xu et al., 2007\)](#page-10-0). We report seasonal data as monthly means  $(n = 1-9$  samples sampled during 1–5 days per month for  $CO_2$  concentrations,  $\delta^{13}C$ ,  $\delta^{18}$ O and  $\Delta_{47}$  and  $n = 1-2$  per month for  $\Delta^{14}$ C). The precision of  $\Delta_{47}$  measurements is estimated as 0.005‰ (standard error,  $n = 28$ ) based on the reproducibility of measurement of cylinder  $CO_2$  working standard  $(\delta^{13}C = -10.7\%)$  $\delta^{18}O = 31.9\%, \Delta_{47} = 0.871\%$  performed between November 2005 and April 2006 or as 0.015‰ (standard error,  $n = 8$ ) based on the reproducibility of measurements of  $CO<sub>2</sub>$  produced by acid digestion of the carbonate standard NBS-19. Precision was slightly lower in the beginning of the seasonal time series. The typical standard error of Pasadena air  $\Delta_{47}$  monthly means is 0.024‰ (n = 1–9).

## 2.2. Calculation of clean air  $\Delta_{47}$

The fraction of  $CO<sub>2</sub>$  from combustion sources was estimated from the observed  $\Delta^{14}$ C as  $f_{ff} = (\Delta^{14}C-60)$ /  $(-960 - 60)$ , where  $-960\%$  is the  $\Delta^{14}$ C value observed in car exhaust and 60% is the  $\Delta^{14}$ C in clean air during 2004–2005 (the estimation assumes that the  $\Delta^{14}$ C associated with non fossil fuel sources is similar to the atmospheric background value). This was used to estimate the  $\Delta_{47}$  values in clean air by removing this contribution from local combustion sources, as  $\Delta_{47\text{-clean}} = (\Delta_{47\text{-observed}} - 0.41f_{ff})/(1$  $f_{\text{ff}}$ ) where 0.41‰ is the  $\Delta_{47}$  value observed in car exhaust ([Affek and Eiler, 2006\)](#page-10-0). This calculation assumes that  $CO<sub>2</sub>$  produced by combustion of natural gas in local power plants, whose  $\Delta_{47}$  value was not yet determined, is similar to that observed in car exhaust. As a sensitivity test for this assumption, we calculated  $\Delta_{47\text{-clean}}$  with a  $\Delta_{47}$  value of zero for both combustion sources. This reduces the seasonal variations in  $\Delta_{47\text{-clean}}$  but does not eliminate them and the values during October to December 2004 remained significantly lower then those during summer 2005.

 $\Delta^{14}$ C data were available for October 2004–December 2005. However, in the summer months (June to October) <span id="page-2-0"></span>of 2005 applying the above correction to estimate clean air  $\Delta_{47}$  values would result in values that are significantly higher than the values reflecting thermodynamic equilibrium, which is unlikely given the current knowledge of the factors controlling  $\Delta_{47}$  values [\(Eiler and Schauble, 2004;](#page-10-0) [Affek and](#page-10-0) [Eiler, 2006\)](#page-10-0). Although a CO<sub>2</sub> source having  $\Delta_{47}$  values significantly higher than those at thermodynamic equilibrium cannot be ruled out, it is more likely that during these months  $\Delta_{47}$  values were modified by fast local exchange with water that eliminated the regional fossil fuel contribution to  $\Delta_{47}$  values but did not affect the  $\Delta^{14}$ C. We therefore consider the clean air  $\Delta_{47}$  values during June to October 2005 to be identical to the uncorrected values, resulting in a minimum estimate for the winter–summer 2005 difference in clean air.

## 3. RESULTS

# 3.1. Seasonal variations

Monthly averaged CO<sub>2</sub> concentrations,  $\delta^{13}C$ ,  $\delta^{18}O$ ,  $\Delta_{47}$ and  $\Delta^{14}$ C values are given in Fig. 1 and [Table 1.](#page-3-0) During the two-year measurement period,  $CO<sub>2</sub>$  concentrations averaged 398  $\pm$  1 µmol mol<sup>-1</sup> (average  $\pm$  SE, n = 101) with no clear seasonal trend, reflecting a contribution of approximately 20  $\mu$ mol mol<sup>-1</sup> from fuel combustion and other sources in the Los Angeles basin.

Values of  $\delta^{13}$ C were relatively low in winter 2004  $(-9.6 \pm 0.2\%)$  and higher in summer  $(-9.2 \pm 0.1\%)$ , in agreement with the seasonal trend observed in northern hemisphere clean, non-urban air—although the Pasadena values are lower at any given time due to local sources (mostly combustion). During 2005, variable contributions of local sources overcame the typical seasonal signal and no clear seasonal trend was observed in  $\delta^{13}$ C. The values of  $\delta^{18}$ O averaged 41.2  $\pm$  0.1‰ over the two years and showed pronounced minima during September 2004  $(40.3 \pm 0.1\%)$  and August 2005 (40.2  $\pm$  0.1‰).

 $\Delta_{47}$  value was lowest (0.75‰) in March 2004, increased to 0.89‰ in May 2004 and then decreased again to 0.80‰ in November.  $\Delta_{47}$  values then increased gradually to the maximal observed value (1.01‰) in August 2005 followed by a decrease to 0.88‰ in December 2005.  $\Delta^{14}$ C values varied between 51‰ in October 2004 and  $-30\%$  in November 2005 with no clear seasonal trend.

## 3.2. Diurnal variations

Diurnal variations in CO<sub>2</sub> concentrations,  $\delta^{13}C$ ,  $\delta^{18}O$ ,  $\Delta_{47}$  and  $\Delta^{14}$ C are given in [Fig. 2](#page-4-0) and [Table 2](#page-5-0). On February



Fig. 1. (a) Seasonal variations (calculated as monthly averages) in  $\Delta_{47}$  (bold circles with full line) and  $\Delta^{14}C$  (empty circles with short-dashed line) for CO<sub>2</sub> in air sampled in Pasadena, CA, during the years 2004 and 2005. The long-dashed line depicts  $\Delta_{47}$  values at thermodynamic equilibrium, based on the monthly mean temperatures in Pasadena, CA. (b) Observed seasonal variation in  $\Delta_{47}$  (bold circles with dashed line) and modelled  $\Delta_{47}$  (full line). (c) Monthly averaged seasonal variations in  $\delta^{13}C$  (observation: empty squares with long-dashed line, model: full line) and (d)  $\delta^{18}O$  (observation: bold squares with short-dashed line, model: full line). Note that the model results are more reliable after October 2004 (where  $\Delta^{14}$ C data is available).

<span id="page-3-0"></span>Table 1

CO<sub>2</sub> concentrations (µmol mol<sup>-1</sup>),  $\delta^{13}C$ ,  $\delta^{18}O$ ,  $\Delta_{47}$  (±SE) and  $\Delta^{14}C$  (%e) observed in air sampled in Pasadena, CA, during the years 2004 and 2005

Month $(n)$	[CO <sub>2</sub> ]	$\delta^{13}C$	$\delta^{18}O$	$\Delta_{47}$	$\Delta^{14} \text{C}$ $%$ Fuel $a$		% Local contribution <sup>b</sup>	
$Jan-04$								
$Feb-04(2)$	$391 \pm 2$	$-9.61 \pm 0.08$	$40.52 \pm 0.17$	$0.784 \pm 0.015$			3.2	
Mar-04 $(2)$	$399 \pm 1$	$-9.94 \pm 0.04$	$40.81 \pm 0.01$	$0.746 \pm 0.028$			5.0	
Apr-04 $(2)$	$404 \pm 0.2$	$-10.10 \pm 0.10$	$41.92 \pm 0.08$	$0.844 \pm 0.112$			5.8	
May-04 $(2)$	$383 \pm 1$	$-9.55 \pm 0.21$	$40.74 \pm 0.02$	$0.890 \pm 0.019$			0.7	
$Jun-04$								
July-04 $(2)$	$399 \pm 2$	$-9.43 \pm 0.23$	$41.45 \pm 0.25$	$0.854 \pm 0.013$			5.3	
Aug-04 $(1)$	403	$-9.03$	41.45	0.877			6.8	
Sept-04 $(6)$	$392 \pm 3$	$-9.19 \pm 0.08$	$40.26 \pm 0.14$	$0.842 \pm 0.027$			4.5	
$Oct-04(3)$	$396 \pm 2$	$-8.94 \pm 0.26$	$41.49 \pm 0.78$	$0.838 \pm 0.006$	51.4	0.8	5.4	
Nov-04 $(8)$	$400 \pm 2$	$-9.41 \pm 0.06$	$41.53 \pm 0.07$	$0.804 \pm 0.020$	$-2.6$	6.1	6.0	
Dec-04 $(8)$	$396 \pm 4$	$-9.20 \pm 0.11$	$41.62 \pm 0.33$	$0.817 \pm 0.017$	8.6	5.0	4.6	
$Jan-05(9)$	$395 \pm 3$	$-9.45 \pm 0.20$	$41.04 \pm 0.33$	$0.880 \pm 0.008$	$-3.7$	6.2	4.1	
Feb $-05(6)$	$400 \pm 3$	$-9.74 \pm 0.15$	$41.22 \pm 0.28$	$0.898 \pm 0.014$	$-11.2$	7.0	4.9	
Mar $-05(9)$	$404 \pm 4$	$-9.87 \pm 0.19$	$41.11 \pm 0.23$	$0.868 \pm 0.012$	$-9.0$	6.8	5.6	
Apr-05 $(9)$	$387 \pm 1$	$-9.26 \pm 0.13$	$41.84 \pm 0.10$	$0.891 \pm 0.013$	17.3	4.2	1.3	
May-05 $(8)$	$400 \pm 5$	$-9.67 \pm 0.23$	$41.42 \pm 0.13$	$0.927 \pm 0.029$	16.9	4.2	4.4	
Jun-05 $(6)$	$398 \pm 4$	$-9.78 \pm 0.26$	$41.03 \pm 0.36$	$0.965 \pm 0.017$	13.1	4.6	3.9	
July-05 $(2)$	$410 \pm 2$	$-10.38 \pm 0.01$	$40.46 \pm 0.01$	$0.951 \pm 0.020$	$-28.6$	8.7	7.1	
Aug-05 $(3)$	$398 \pm 1$	$-9.59 \pm 0.03$	$40.23 \pm 0.06$	$1.008 \pm 0.060$	$-10.0$	6.9	4.7	
Sept-05 $(2)$	$396 \pm 2$	$-9.68 \pm 0.09$	$41.27 \pm 0.06$	$0.991 \pm 0.028$			4.8	
$Oct-05(3)$	$389 \pm 2$	$-9.29 \pm 0.01$	$41.18 \pm 0.01$	$0.943 \pm 0.001$	7.6	5.1	3.0	
Nov-05 $(5)$	$420 \pm 2$	$-10.71 \pm 0.06$	$40.97 \pm 0.07$	$0.878 \pm 0.014$	$-29.8$	8.8	9.8	
Dec-05 $(2)$	$403 \pm 2$	$-10.02 \pm 0.06$	$40.87 \pm 0.05$	$0.876 \pm 0.020$	$-8.4$	6.7	5.7	
Monthly mean SE	2	0.12	0.17	0.024				
Car exhaust sample		$-25.03$	30.37	0.361	$-959$			

Data is given as monthly averages, with the number of samples per month  $n = 1-9$  per month over 1–5 days per month.

<sup>a</sup> % Fuel refers to the amount of CO<sub>2</sub> in Pasadena air originating from fossil fuel combustion, based on  $\Delta^{14}$ C data and calculated as: % fuel =  $(\Delta^{14}C-60)/(-960-60)$  100.

<sup>b</sup> % Local contribution refers the net effect of local CO<sub>2</sub> fluxes, estimated based on the CO<sub>2</sub> concentration difference between Pasadena, CA and Mauna Loa, HI (Mauna Loa data from the NOAA CMDL flask network, [http://www.esrl.noaa.gov/gmd/dv/ftpdata.html\)](http://www.esrl.noaa.gov/gmd/dv/ftpdata.html)).

14th,  $CO<sub>2</sub>$  concentrations varied from 499 µmol mol<sup>-1</sup> at 5am to 398  $\mu$ mol mol<sup>-1</sup> between 8 and 11pm.  $\delta^{13}$ C values were a mirror image of the concentration trend, varying between  $-9.0$  and  $-13.0%$  with a 'Keeling plot' intercept of  $-29.4 \pm 0.5\%$  ( $R^2 = 0.996$ ).  $\Delta_{47}$  values varied in a similar fashion from  $0.818 \pm 0.004\%$  between 2am and 2pm and  $0.87 \pm 0.01\%$  between 5pm and 11pm. The  $\Delta^{14}$ C values were highly correlated with the  $\delta^{13}$ C values  $(\delta^{13}C = (0.034 \pm 0.001)\Delta^{14}C - (10.16 \pm 0.06);$   $R^2 = 0.992)$ and varied between  $-80\%$  at 5am and 41‰ at 11pm. The  $\Delta^{14}$ C 'Keeling plot' intercept was  $-569\%$  ( $R^2 = 0.986$ ).

On February 21st,  $CO<sub>2</sub>$  concentrations were high at night and low around midday, as is typically observed in the region [\(Newman et al., 2004](#page-10-0)). They varied between 453 µmol mol<sup>-1</sup> at 6am and 392 µmol mol<sup>-1</sup> at 3pm.  $\delta^{13}C$ values varied between  $-8.7$  and  $-11.7%$  with a 'Keeling plot' whose intercept was  $-33 \pm 1\%$  ( $R^2 = 0.978$ ).  $\Delta_{47}$  values were high  $(0.88 \pm 0.02\%)$  between noon and 6pm and low  $(0.80 \pm 0.01\%)$  in early morning and late evening. The  $\Delta^{14}$ C trend was a mirror image of the CO<sub>2</sub> concentrations and varied from  $-50\%$  at 6am to 47‰ at 3pm with a 'Keeling plot' intercept of  $-638\%$  ( $R^2 = 0.962$ ). The relationship between  $\delta^{13}$ C and  $\Delta^{14}$ C was similar to the other two days, though the correlation was not as clear  $(\delta^{13}C = (0.019 \pm 0.009)\Delta^{14}C$ - $(9.8 \pm 0.3); R^2 = 0.480).$ 

On March 24th,  $CO<sub>2</sub>$  concentrations were high at night and low around midday, between  $467 \mu$ mol mol $^{-1}$  at 5am and 406 µmol mol<sup>-1</sup> at 11am.  $\delta^{13}$ C values varied between  $-9.6$  and  $-11.8\%$  with a 'Keeling plot' intercept of  $-28 \pm 1\%$  ( $R^2 = 0.985$ ). The  $\Delta_{47}$  pattern was generally similar to that of February 21st though with a less clear trend and varied between 0.73‰ (at 8am) and 0.93‰ (at 11am). The  $\Delta^{14}$ C trend was a mirror image of CO<sub>2</sub> concentrations and varied between 57‰ at 8am and 17‰ at 2am with a 'Keeling plot' intercept of  $-660\%$  ( $R^2 = 0.922$ ). Correlation was observed between  $\delta^{13}$ C and  $\Delta^{14}$ C  $( \delta^{13}C=(0.024 \pm 0.004) \Delta^{14}C-(10.2 \pm 0.1);$   $R^2=0.822$ ). In all 3 days the  $\delta^{18}$ O diurnal pattern was similar to that of  $\delta^{13}$ C but the correlation with concentrations was weaker (a 'Keeling plot' correlation coefficient of 0.503 for all 3 days, as compared to  $R^2 = 0.975$  for  $\delta^{13}$ C).

The different daily pattern among the three days was probably due to different weather patterns that reflect potential variations in the trajectories of the air observed. February 14th was a clear day that turned cloudy between 4 and 5pm. The minimal and maximal temperatures were 9 and 21  $\degree$ C, respectively. Early morning (before 8am) was characterized by light North-westerly wind (wind data is based on measurements in a private station located at 34.136N, 118.124W, and reported in the Weather Under-

<span id="page-4-0"></span>

Fig. 2. (a) Diurnal variations in  $\Delta_{47}$  (bold circles with full line) and  $\Delta^{14}C$  (empty circles with short-dashed line) in air sampled on February 14, February 21 and March 24, 2006 in Pasadena, CA. The long-dashed line depicts  $\Delta_{47}$  values estimated for "corrected" clean air. (b) Diurnal variations in CO<sub>2</sub> concentrations (grey circles with full line),  $\delta^{13}C$  (empty squares with long-dashed line), and  $\delta^{18}O$  (bold squares with shortdashed line).

ground web page: [http://www.wunderground.com/weath](http://www.wunderground.com/weatherstation/WXDailyHistory.asp?ID=KCAPASAD5)[erstation/WXDailyHistory.asp?ID=KCAPASAD5\)](http://www.wunderground.com/weatherstation/WXDailyHistory.asp?ID=KCAPASAD5). After 8am slightly stronger wind blew from the Southeast. February 21st was a clear and chilly day with minimal and maximal temperatures of 3 and  $18 \degree C$ , respectively. Early morning (before 8am) wind direction was North-westerly, followed by 3 h of South easterly winds. From midday till midnight winds were stronger with alternating direction between Northeast and Northwest. March 24th was partly cloudy with minimal and maximal temperatures of 8 and 25  $\degree$ C, respectively. Early morning (before 8am) and late night (after 8pm) wind direction was North-westerly. Winds intensified during the day and changed gradually from South easterly before noon and South westerly after noon. Sunrise occurred between 5:50 and 6:20am. Sunset occurred between 5:35 and 6:10 pm.

#### 4. DISCUSSION

#### 4.1. Seasonal variations

Previous interpretations of the  $\delta^{18}$ O value of atmospheric  $CO<sub>2</sub>$  have emphasized the dominant influence of fluxes that involve exchange equilibria between  $CO<sub>2</sub>$  and ocean water, leaf water and soil water. This interpretation implies, a priori, that the  $\Delta_{47}$  value of these CO<sub>2</sub> fluxes should also reflect thermodynamic equilibrium at the temperature of that exchange because a system in equilibrium with respect to  $CO<sub>2</sub>–H<sub>2</sub>O$  exchange will also be in equilibrium with respect to homogeneous equilibria among  $CO<sub>2</sub>$ isotopologues (as for  $\delta^{18}O$  ([Gillon and Yakir, 2000](#page-10-0)), approximately 15 hydration steps are necessary for full exchange that would result in equilibrium  $\Delta_{47}$  values). At thermodynamic equilibrium, the value of  $\Delta_{47}$  varies due

<span id="page-5-0"></span>Table 2

CO<sub>2</sub> concentrations (µmol mol<sup>-1</sup>),  $\delta^{13}$ C,  $\delta^{18}$ O,  $\Delta_{47}$ , and  $\Delta^{14}$ C (‰) observed in air sampled in Pasadena, CA, over a 24 h cycle during February 14th, February 21st, and March 24th, 2006

Time	$[CO_2]$	$\delta^{13}C$	$\delta^{18}O$	$\Delta_{47}$	$\Lambda^{14} \text{C}$
February, 14					
2:00	479	$-12.57$	40.41	0.834	$-71.0$
5:00	499	$-13.01$	40.80	0.815	$-80.1$
8:00	435	$-10.67$	41.56	0.813	$-16.3$
11:00	441	$-11.07$	40.96	0.814	$-34.3$
14:00	406	$-9.25$	41.83	0.814	25.9
17:00	401	$-9.14$	42.26	0.880	26.5
20:00	398	$-9.09$	42.25	0.845	34.6
23:00	398	$-8.97$	42.25	0.870	41.2
February, 21					
3:00	441	$-11.68$	40.80	0.953	$-35.5$
6:00	453	$-11.87$	40.73	0.786	$-50.0$
9:00	409	$-9.64$	41.08	0.792	15.2
12:00	402	$-9.44$	41.24	0.888	16.1
15:00	392	$-8.69$	41.57	0.846	47.0
18:00	403	$-9.32$	41.12	0.910	
21:00	422	$-10.48$	40.92	0.830	$-16.1$
23:59	435	$-10.85$	41.15	0.804	$-19.9$
March, 24					
2:00	418	$-10.22$	41.63	0.872	17.1
5:00	467				
8:00	457	$-11.78$	41.05	0.793	$-57.1$
11:00	406	$-9.74$	41.44	0.932	11.1
14:00	424	$-10.41$	40.84	0.822	$-19.6$
17:00	409	$-9.62$	41.29	0.828	15.9
20:00	427	$-10.59$	40.92	0.844	$-17.1$
23:00	457	$-11.62$	40.24	0.807	$-56.7$

to a temperature-dependent tendency to 'clump' <sup>13</sup>C and <sup>18</sup>O into bonds with each other leading to higher  $\Delta_{47}$  values at low equilibrium temperatures ([Eiler and Schauble, 2004;](#page-10-0) [Wang et al., 2004](#page-10-0)). Under such conditions, the global mean  $\Delta_{47}$  value should reflect a flux-weighted average of global temperatures. Local values are likely to be a mixture of background air values with  $\Delta_{47}$  values of local fluxes that in turn would reflect the local temperatures of exchange. In the extreme case in which all the  $CO<sub>2</sub>$  undergoes exchange locally, the predicted seasonal variations, based on long-term monthly mean temperatures in Pasadena, CA, would be between a winter maximum  $\Delta_{47}$  value of 0.99%  $(T = 13 \text{ °C})$  and a summer minimum of 0.92‰  $(T = 25 \text{ °C}$ ; [Fig. 1](#page-2-0)a). Under more realistic conditions, in which only part of the  $CO<sub>2</sub>$  interacts with local water, seasonal variations in  $\Delta_{47}$  are expected to be even smaller.

Monthly mean  $\Delta_{47}$  values of atmospheric CO<sub>2</sub> in Pasadena, CA, exhibit a temporal variation [\(Fig. 1a](#page-2-0); [Table 1\)](#page-3-0) that has pronounced seasonality but is different in three respects from that predicted for thermodynamic equilibrium: The observed values are lower in winter and higher in summer—opposite to the predicted seasonal trend; the seasonal amplitude is larger than predicted; and there is a pronounced, unexpected inter-annual trend. This seasonal pattern and inter-annual trend differs from those for  $\delta^{13}$ C and  $\delta^{18}$ O ([Fig. 1c](#page-2-0) and d).

Our results suggest that seasonal variations are the result of variations in the balance between several end-members,

some of which have  $\Delta_{47}$  values close to thermodynamic equilibrium, resulting from full  $CO<sub>2</sub>–H<sub>2</sub>O$  exchange, or even slightly above the predicted equilibrium values, and at least one flux that has a  $\Delta_{47}$  value substantially lower than the equilibrium value. Furthermore,  $\Delta_{47}$  values are closest to predicted equilibrium values during the summer and furthest away during winter, suggesting that the fluxes that dominate the summer budget reflect isotopic equilibrium (presumably mediated by exchange with water) or possibly have  $\Delta_{47}$  values slightly above that at equilibrium, whereas one or more of the local fluxes important to the winter budget have low  $\Delta_{47}$  values.

We would therefore expect the  $\Delta_{47}$  values of CO<sub>2</sub> in the major fluxes to and from the atmosphere (respiration, photosynthesis and ocean gas exchange) to be between ca. 0.9 and 1.0‰ [\(Eiler and Schauble, 2004; Wang et al., 2004\)](#page-10-0). Such values are consistent with the summer maximum observed in our results, but are inconsistent with the lower values observed in winter. One possible explanation is that the low  $\Delta_{47}$  values characteristic of winter Pasadena air reflect a greater proportion of  $CO<sub>2</sub>$  from automobile exhaust and other combustion sources, which are known to have average  $\Delta_{47}$  values near 0.41‰ ([Affek and Eiler, 2006\)](#page-10-0). The extent of local transportation is not likely to vary strongly with season and emissions from local power generation is significantly higher in summer than in winter (E. Klinkner, Pasadena Water and Power, personal communication, 2006). However, relative anthropogenic contribution might be higher if the mixed layer is shallower in winter or if the trajectory of the air reaching Pasadena in winter leads to increased urban contribution. We tested this hypothesis by analyzing some of our air samples for  $\Delta^{14}$ C ([Fig. 1a](#page-2-0); [Table 1](#page-3-0)), which can identify  $CO<sub>2</sub>$  produced from fossil fuel combustion ([Zondervan and Meijer, 1996; Levin](#page-10-0) [et al., 2003; Turnbull et al., 2006\)](#page-10-0). Correlation between the  $\Delta^{14}$ C and  $\Delta_{47}$  values over the seasonal cycle (primarily through the winter) would suggest that anthropogenic emissions control the seasonal variation in  $\Delta_{47}$  values and are the cause for the low  $\Delta_{47}$  values observed in winter. CO<sub>2</sub> in clean air has a  $\Delta^{14}$ C value of ca. 60‰ (i.e., the value of 65‰ reported for the beginning of 2004 ([Turnbull et al.,](#page-10-0) [2006\)](#page-10-0) decreased by the mean rate of  $\sim$ 5‰/y ([Levin et al.,](#page-10-0) [2003\)](#page-10-0)). CO<sub>2</sub> from local car exhaust has a  $\Delta^{14}$ C value of  $-960\%$  ([Table 1,](#page-3-0) the value is not  $-1000\%$  probably due to a few percent of ethanol, having modern  $\Delta^{14}$ C values, added to the gasoline). During the period of October 2004 to December 2005, the  $\Delta^{14}$ C values of CO<sub>2</sub> in Pasadena air varied between  $51\%$  and  $-30\%$ . This suggests that the contribution of  $CO<sub>2</sub>$  from fossil fuel combustion varied between 1% and 9%, averaging 6% ([Table 1](#page-3-0)). No seasonal trend was observed in the  $\Delta^{14}$ C values, suggesting that there is no strong seasonality in the anthropogenic  $CO<sub>2</sub>$  contribution in Pasadena and hence probably no significant seasonal shift in the air trajectory that would lead to more urban air arriving to Pasadena in either season. No correlation was observed between the  $\Delta^{14}$ C and  $\Delta_{47}$  values ([Fig. 4\)](#page-6-0) indicating that fossil fuel combustion is not a major control on the seasonal trend in  $\Delta_{47}$  values.

We can use the  $\Delta^{14}$ C data to "correct" the measured  $\Delta_{47}$ values for the contribution from local combustion sources

<span id="page-6-0"></span>

Fig. 3. Correlation between  $\Delta_{47}$  and  $\Delta^{14}$ C values observed in samples collected as part of our study of diurnal variations. The full line is a fit to the combined data of all 3 days (excluding 2 outliers) and yields the line:  $\Delta_{47} = (0.0004 \pm 0.0001)\Delta^{14}C +$ (0. 834  $\pm$  0.006), which implies  $\Delta_{47} = 0.86 \pm 0.03\%$  for clean air (based on  $\Delta^{14}C = 60\%$  in clean air). This line crosses the observed composition of car exhaust ( $\Delta_{47} = 0.41\%$ ,  $\Delta^{14}$ C = -960‰). The dashed line reflects a mixing line between the measured composition of car exhaust and a hypothetical composition of clean air if it were at thermodynamic equilibrium (0.98‰ for a monthly mean local temperature of 15  $\degree$ C in February and March). The error bars are the standard error in 5 measurement of each sample (approximately 0.02‰), the error in  $\Delta^{14}$ C is estimated as  $\sim$ 2‰.



Fig. 4. Lack of correlation between  $\Delta_{47}$  and  $\Delta^{14}$ C measured between October 2004 and December 2005.

in order to estimate  $\Delta_{47}$  values of nominally clean local air. This clean air is not identical to background atmosphere, as it is air free of local fossil fuel emissions but not free of the contribution of other local fluxes, such as photosynthesis and respiration by local ecosystems. The resulting  $\Delta_{47}$  values for clean air are higher by 0.02‰ to 0.05‰ than the measured values, yielding an average  $\Delta_{47}$  value of 0.94‰ for 2005 ([Fig. 1](#page-2-0)b). The correction does not substantially change the seasonal trend, suggesting that local anthropogenic CO<sub>2</sub> sources affect the mean  $\Delta_{47}$  value of Pasadena air but have only a minor effect on the temporal variations and are not the cause of the seasonal cycle. The fossil fuel contribution derived from  $\Delta^{14}$ C is similar to the local contribution to  $CO<sub>2</sub>$  concentration (5% in average; [Table 1](#page-3-0)) estimated based on the difference of monthly mean concentrations between Pasadena and Mauna Loa (data from the NOAA/ESRL flask network, [http://www.esrl.noaa.gov/](http://www.esrl.noaa.gov/gmd/dv/ftpdata.html) [gmd/dv/ftpdata.html](http://www.esrl.noaa.gov/gmd/dv/ftpdata.html)) indicating that the net contribution of the local ecosystem is small. However,  $\Delta_{47}$ , just like  $\delta^{18}$ O, is likely to be affected by gross fluxes and therefore might be influenced by local processes far more than suggested by concentration variations alone.

The combined constraints of  $\Delta^{14}$ C and  $\Delta_{47}$  require that one or more of the CO<sub>2</sub> sources other than anthropogenic emissions have a  $\Delta_{47}$  value lower than expected for thermodynamic equilibrium over the range of local temperatures. Our results are most consistent with this low- $\Delta_{47}$  source being dominant in winter; based on previous studies of  $\Delta_{47}$  in CO<sub>2</sub> from natural and anthropogenic sources [\(Eiler](#page-10-0) [and Schauble, 2004; Affek and Eiler, 2006](#page-10-0)), this source must also make up a large fraction of all fluxes to the atmosphere because no known flux differs in  $\Delta_{47}$  from average atmospheric  $CO<sub>2</sub>$  by more than ca. 0.5‰. Ecosystem respiration (both plants and soil) is the most obvious source of this  $CO<sub>2</sub>$ , though ocean–air exchange cannot be ruled out *a priori*. If so, atmospheric  $CO<sub>2</sub>$  would be dominantly influenced by respiration (or ocean exchange) in winter, and drawn towards equilibrium with leaf water between late spring and early fall, when photosynthesis dominates.

To assess the possibility that the low  $\Delta_{47}$  values of winter background air might reflect a possible kinetic isotope effect associated with atmosphere-ocean exchange (whose  $\Delta_{47}$  values have not yet been characterized), we examined  $CO<sub>2</sub>$  in air from Cape Grim, Tasmania, and Barrow, Alaska. Both sites are considered remote enough from anthropogenic point sources so that air is likely to reflect global (or hemispheric) background air, not affected by local anthropogenic sources. Air in Cape Grim was sampled when the wind direction was west to south-west, suggesting probable marine influence. If  $\Delta_{47}$  values of global background air or in ocean–air exchange were significantly lower than equilibrium values, we would expect these air samples to reflect these low values. However,  $\Delta_{47}$  values of marine air collected at Cape Grim and Barrow during 2005 averaged  $0.92 \pm 0.01\%$  ([Table 3](#page-7-0)), higher than the values we observed in winter-time Pasadena air, not supporting this hypothesis. Thus, respiration seems like the most plausible source of low  $\Delta_{47}$  values in nominally clean air.

# 4.2. Diurnal variations

We further explored the isotopic composition of respired  $CO<sub>2</sub>$  by examining the diurnal variations in  $\Delta_{47}$  in Pasadena air, expecting that respired  $CO<sub>2</sub>$  would be dominant at night and therefore lead to lower night-time  $\Delta_{47}$  values (note that due to the high sensitivity of  $\Delta_{47}$  to even small amounts of water, direct sampling of soil or leaf respired  $CO<sub>2</sub>$  using conventional incubation methods is problematic, forcing us to use this indirect approach). This approach does not distinguish between soil and dark plant respiration. Measurements of this kind are likely to be influenced by variations in the relative contribution of all local sources, including anthropogenic ones, due to diurnal

<span id="page-7-0"></span>Table 3

Isotopic composition of air sampled in Cape Grim, Tasmania				
(40.68S, 144.68E) and Barrow, AK (71.32N, 156.6W)				



variations in atmospheric mixing as a result of the typically shallow mixed layer at night (analogous to the ''rectifier effect''; [Denning et al., 1996\)](#page-10-0). Similar effects have been previously observed in diurnal measurements of urban air and were manifested as a lack of correlation between source strength and  $CO_2$  concentrations and  $\delta^{13}C$  values near the surface ([Reid and Steyn, 1997; Grimmond et al., 2002; Ne](#page-10-0)[mitz et al., 2002; Pataki et al., 2005; Moriwaki et al., 2006\)](#page-10-0). To control for anthropogenic contributions and normalize the effect of varying synoptic conditions and mixed layer depth, we compared  $\Delta_{47}$  diurnal variations to those in  $\Delta^{14}$ C (since those physical atmospheric parameters would control both tracers in a similar fashion).

The values of  $\Delta_{47}$  and  $\Delta^{14}$ C vary inversely with concentrations over a 24-h cycle for 3 separate days during winter 2006 ([Fig. 2](#page-4-0) and [Table 2](#page-5-0)). The values of  $\Delta_{47}$  and  $\Delta^{14}$ C are generally correlated with one another [\(Fig. 3](#page-6-0); excepting the data for air collected at 3am on February 21st and 11am on March 24th). A likely explanation of this correlation is diurnal variations in atmospheric mixing, so that  $CO<sub>2</sub>$  from both fossil fuel combustion and respiration is concentrated near the surface at night and is mixed deeper in the atmosphere during the day.

The concentration pattern during February 21st and March 24th is the pattern observed typically in Pasadena ([Newman et al., 2004\)](#page-10-0). It is associated with the typical wind pattern in the Los Angeles basin with wind blowing seawards (Northerly wind) in early morning and blowing from sea to inland in the afternoon ([Neiburger, 1969](#page-10-0)) and returning to seaward direction in the evening. In February 14th, on the other hand, sea breeze persisted through the evening and night, leading to the observed evening cloud cover, which is typically accompanied with low  $CO<sub>2</sub>$  concentrations (S. Newman, personal communication, 2007). Although the sea breeze is likely to bring to Pasadena urban air from Los Angeles this wind direction is not necessarily associated with higher concentrations and larger anthropogenic contribution, suggesting that the boundary layer height and local surface fluxes are more important than wind direction to the local Pasadena  $CO<sub>2</sub>$  diurnal pattern.

The correlation in [Fig. 3](#page-6-0) can be fit with a line that crosses the observed car exhaust  $\Delta^{14}$ C value of -960‰ at a  $\Delta_{47}$  value of 0.41‰—identical to the  $\Delta_{47}$  value measured in car exhaust in the Pasadena area [\(Affek and Eiler, 2006\)](#page-10-0). This result strongly indicates that one cause of diurnal variations in  $\Delta^{14}C$ ,  $\Delta_{47}$  and  $\delta^{13}C$  is concentration of car exhaust in boundary layer air at night. On the other hand, given a  $\Delta^{14}$ C value of 60‰ for clean air, this line suggests a  $\Delta_{47}$  value of  $0.86 \pm 0.03\%$  for clean air (i.e., not including the local fossil fuel component). This value is significantly lower than that at thermodynamic equilibrium (0.98‰ for the mean local temperature in February and March) indicating, as discussed above for the seasonal variations, that the clean air end member does not reflect equilibrium and that at least one of the non-fossil-fuel  $CO<sub>2</sub>$  fluxes driving diurnal variability is characterized by lower  $\Delta_{47}$  values.

As with the seasonal data, we used  $\Delta^{14}$ C values to "correct" the measured  $\Delta_{47}$  values for the contribution of combustion sources in order to estimate diurnal variations of  $\Delta_{47}$  from other sources. The "corrected" values are all higher than the raw measured values, but the pattern of diurnal variability changed only slightly [\(Fig. 2](#page-4-0)). One explanation of this result is that the shallow mixing also concentrates respired  $CO<sub>2</sub>$  into boundary layer air at night, and that local respired  $CO_2$  is characterized by a  $\Delta_{47}$  value lower than expected for thermodynamic equilibrium. This interpretation is consistent with our suggested interpretation of the seasonal cycle; i.e., competition between low  $\Delta_{47}$  values in respiration and higher  $\Delta_{47}$  values resulting from CO<sub>2</sub>-H<sub>2</sub>O (leaf or surface) exchange are responsible for seasonal variations in  $\Delta_{47}$ , and the diurnal cycle in urban air reflects accumulation of both low- $\Delta_{47}$  respiration CO<sub>2</sub> and very low- $\Delta_{47}$  combustion CO<sub>2</sub> in night-time boundary layer air.

We estimate the  $\Delta_{47}$  and  $\Delta^{14}$ C values for the respiration component that we hypothesize contributes to diurnal variability in Pasadena air using the observed, uncorrected, diurnal data in a 'Keeling plot' approach [\(Keeling, 1958\)](#page-10-0). The intercept of the  $\Delta^{14}C$  'Keeling plot' is  $-595 \pm 29\%$ , indicating that the source of night-time  $CO<sub>2</sub>$  is a combination of  $64\%$  combustion  $CO<sub>2</sub>$  and  $36\%$  modern carbon, which we attribute to respiration. The  $\Delta_{47}$  'Keeling plot' intercept is  $0.54 \pm 0.10\%$ , implying a  $\Delta_{47}$  value of  $0.77 \pm 0.28\%$  for the putative respiration end-member. The error estimate for the respiration  $\Delta_{47}$  value is large, probably because of the indirect approach we had to take in estimating it through diurnal variations as well as due to the complication in using urban air. Future work must be devoted to detailed respiration measurements in a clean environment, such as agricultural settings, in order to get a

more precise estimate of the  $\Delta_{47}$  value associated with respiration. Nevertheless, this value is indistinguishable from the value of 0.76‰ observed for human breath [\(Affek and](#page-10-0) [Eiler, 2006](#page-10-0)). Human breath obviously constitutes a trivial fraction of all respired  $CO<sub>2</sub>$ , but it is possible that similar biological kinetic isotope effects are responsible for the lower than equilibrium value in both types of respiration. At this stage we can only hypothesize as to the cause for this low respiration value. This value is clearly out of equilibrium for  $CO<sub>2</sub>$  but might be the result of a kinetic isotope effect either in the  $CO<sub>2</sub>$  production or in its escape from the leaf or the soil. This might be combined with a possibility that the respired CO<sub>2</sub> still carries some of the original  $\Delta_{47}$ signature of the carboxyl group from which it was formed (W. Guo, personal communication, 2007), which due to the lower bond order (as compared to  $CO<sub>2</sub>$ ) is expected to have a lower  $\Delta_{47}$  value at equilibrium ([Schauble et al.,](#page-10-0) [2006](#page-10-0)).

### 4.3. Mass balance model

We can consider the plausibility and implications of the apparent  $\Delta_{47}$  value of respired CO<sub>2</sub> by examining its ability to explain the observed seasonal cycle, using a single box model. We used the isoflux mass-balance approach ([Tans](#page-10-0) [et al., 1993](#page-10-0)), with the flux rates constrained by our Pasadena air CO<sub>2</sub> concentration,  $\delta^{13}$ C and  $\Delta^{14}$ C measurements, with concentrations and  $\delta^{13}$ C values measured in Mauna Loa (data from the NOAA/ESRL flask network, [http://](http://www.esrl.noaa.gov/gmd/dv/ftpdata.html)) [www.esrl.noaa.gov/gmd/dv/ftpdata.html\)](http://www.esrl.noaa.gov/gmd/dv/ftpdata.html)) used as background air values. We then used estimates of  $\Delta_{47}$  values associated with  $CO<sub>2</sub>$  sources to reconstruct the observed  $\Delta_{47}$  seasonal variations.

The mass balance relations used in this model are:

$$
M(c_{\rm a} - c_{\rm b}) = F_{\rm ff} + R + F_{\rm la} - F_{\rm al}
$$
  
\n
$$
M(c_{\rm a}\delta_{\rm a} - c_{\rm b}\delta_{\rm b}) = F_{\rm ff}\delta_{\rm ff} + R\delta_{\rm R} + F_{\rm la}\delta_{\rm la} - F_{\rm al}(\delta_{\rm b} - \varepsilon)
$$
\n(3)

where  $c_a$  and  $c_b$  are  $CO_2$  concentrations in ambient Pasadena air and in background (Mauna Loa) air, and  $\delta_a$  and  $\delta_b$  are the respective isotopic compositions (referring to either  $\delta^{13}C$ ,  $\delta^{18}O$ ,  $\Delta^{14}C$  or  $\Delta_{47}$ ); M is the mass of air in the footprint covered by the observations, namely the size of the box in the model, so that  $Mc_a$  is the pool size of  $CO<sub>2</sub>$  in the box (we have no way to estimate the size of M in the local footprint, therefore all calculated flux rates are normalized to the box size, namely given as F/M, see below);  $F_{\text{ff}}$  is the flux of  $CO_2$  from local fossil fuel combustion,  $R$  is ecosystem respiration flux,  $F_{al}$  and  $F_{la}$  are fluxes of leaf gas-exchange (from air to leaf and from leaf to air, respectively).  $\delta_i$  (or  $\Delta_i$ ) values are the isotopic compositions associated with the respective fluxes, and  $\varepsilon$  is the isotopic fractionation associated with  $CO<sub>2</sub>$  uptake in photosynthesis (taken as 18‰ [\(Bowling et al., 2001\)](#page-10-0) for  $\delta^{13}$ C and zero for the other isotopic tracers).

First, we constructed a steady-state single box model of the annual budget based on isoflux mass balance equations (Eqs. (2, 3)) for  $\Delta_{47}$ , assuming the ratio between the local flux rates to be the same as in global annual flux rates. We consider that  $F_{1a} = A_n(c_c/(c_a - c_c))$ , where  $A_n$  is net photosynthesis rate,  $c_a$  is atmospheric CO<sub>2</sub> concentration and  $c_c$  is the  $CO_2$  concentration in the chloroplasts, and that  $c_c/c_a$ is typically 2/3, yielding:  $F_{1a} = 2A_n$  and  $F_{a1} = 3A_n$ , and  $R = A_n$  and we assume that  $F_{\text{ff}} = 0.1 A_n$ . For  $\Delta_{47}$  we adopt the value  $\Delta_{47-\text{ff}} = 0.41\%$  [\(Affek and Eiler, 2006](#page-10-0)), and  $\Delta_{47-R} = 0.77\%$  (based on diurnal variation analysis above) and  $\Delta_{47-b} = 0.92\%$  (based on values observed in Cape Grim and Barrow air). We assume that ocean exchange fluxes are not involved directly in the local Pasadena air mass balance, but only affect it through possible variations in background air  $\Delta_{47}$  values. We further assume that  $CO_2$  uptake in photosynthesis  $(F_{a})$  does not include a  $\Delta_{47}$  isotopic effect  $(\varepsilon = 0$  in Eq. (3)). Given these assumptions we calculate a  $\Delta_{47}$  value of 0.97‰ for the flux out of leaves ( $F_{1a}$ ). This value is equivalent to an equilibration temperature of 16  $\degree$ C, very close to the mean yearly local temperature of 18  $\degree$ C, and it provides the high  $\Delta_{47}$  end-member of the local  $\Delta_{47}$ balance.

We can use these end-member  $\Delta_{47}$  values to reconstruct the observed seasonal cycle. We first used the concentration and  $\Delta^{14}$ C records to estimate flux rates that would fit the observed  $\delta^{13}$ C seasonal record in Pasadena. We then use the obtained flux rates to fit the observed Pasadena air  $\delta^{18}$ O and  $\Delta_{47}$  trends.

We use  $\Delta^{14}$ C data to estimate the fossil fuel local contribution as:

$$
\frac{F_{\rm ff}}{M} = \frac{\Delta^{14}C_{\rm a} - \Delta^{14}C_{\rm b}}{\Delta^{14}C_{\rm ff} - \Delta^{14}C_{\rm b}} \cdot c_{\rm b}
$$
(4)

using  $\Delta^{14}C_{ff} = -960\%$  and  $\Delta^{14}C_{b} = 60\%$  ([Levin et al.,](#page-10-0) [2003; Turnbull et al., 2006](#page-10-0)) and assuming that  $\Delta^{14}C_R \approx \Delta^{14}C_{1a} \approx \Delta^{14}C_b$ . We then assume that the concentration excess ( $C_E = c_a - c_b$ ) in Pasadena with respect to Mauna Loa reflects the sum of local anthropogenic and biogenic fluxes, so that the difference between that excess and  $F_{\text{ff}}/M$  is the effect of the net local ecosystem flux (with  $A_n/M = 10F_{\rm ff}/M$  and  $R/M=10F_{\rm ff}/M + C_{\rm E}$ ). We then use  $\delta^{13}C_{ff} = -31\%$ , (based on a weighted average of  $\delta^{13}C$  values for the fossil fuel sources used in California, [Newman](#page-10-0) [et al., 2004\)](#page-10-0),  $\delta^{13}C_{1a} = \delta^{13}C_b + \varepsilon$  (with  $\varepsilon = 18\%$ ) and  $\delta^{13}C_R = -26.5\%$  to reconstruct the  $\delta^{13}C$  seasonal variations in Pasadena. In February to September 2004, where  $\Delta^{14}$ C data is not available, we vary  $F_{\text{ff}}/M$  to fit the observed  $\delta^{13}$ C values ([Fig. 1](#page-2-0)c).

The resulting flux rates are given in [Table 4.](#page-9-0) In 2004 the fossil fuel local contribution seem to be high in winter and low in summer (note, however, that in the absence of  $\Delta^{14}$ C data till October 2004, our analysis is more reliable for 2005). The local anthropogenic influence on  $\delta^{13}$ C was therefore in phase with the global  $\delta^{13}$ C variations in background air, resulting in a seasonal pattern that is similar to the global pattern, though shifted to more depleted values. In 2005 the average fossil fuel contribution was the same as in 2004 but it was distributed rather evenly through the year, resulting in the observed absence of a clear seasonal pattern in  $\delta^{13}$ C. During 2005 the fossil fuel component was slightly higher than the  $CO<sub>2</sub>$  concentration excess when comparing Pasadena concentrations to the corresponding Mauna Loa concentrations, suggesting net  $CO<sub>2</sub>$  uptake in photosynthesis year round. In 2004, on the other hand, net respiration flux

<span id="page-9-0"></span>Table 4 Parameters and results of mass balance box model

Month	$F_{\text{ff}}/M$	R/M	$F_{\rm al}/M$	$F_{\rm la}/M$	$\Delta_{47-b}$
Jan-04					
Feb-04	40	373	1200	800	0.92
$Mar-04$	40	380	1200	800	0.92
Apr-04	40	383	1200	800	0.92
$May-04$	60	543	1800	1200	0.92
Jun-04					
July-04	10	111	300	200	0.92
Aug-04	$\overline{2}$	45	60	40	0.92
Sept-04	10	108	300	200	0.92
$Oct-04$	3	51	100	67	0.92
$Nov-04$	25	245	737	491	0.92
Dec-04	20	199	602	401	0.92
$Jan-05$	25	243	756	504	0.92
Feb-05	28	275	852	568	0.92
Mar-05	28	271	829	553	0.92
Apr-05	17	157	507	338	0.92
$May-05$	17	169	505	337	0.85
$Jun-05$	18	182	554	369	0.82
July-05	36	353	1080	720	0.82
Aug-05	28	269	835	557	0.8
Sept-05	20	197	594	396	0.8
$Oct-05$	20	196	615	410	0.82
$Nov-05$	37	371	1101	734	0.88
$Dec-05$	27	269	818	546	0.88
	$\delta^{13}$ C end-members		$-26.5$	$\delta^{13}C_b - 18$	$\delta^{13}C_b - 18$
$\delta^{18}$ O end-members		30	31.5	$\delta^{18}O_b$	47
$\Delta^{14}$ C end-members		$-960$	60	60	60
$\Delta_{47}$ end-members		0.41	0.77	$\Delta$ <sub>47-b</sub>	0.97

Flux rates are normalized to the air pool size and are given in units of  $\mu$ mol mol<sup>-1</sup>. Isotopic values are given in units of ‰. Concentrations,  $\delta^{13}$ C, and  $\delta^{18}$ O in background air values are taken as the values observed in Mauna Loa (of the NOAA CMDL flask network, [http://www.esrl.noaa.gov/gmd/dv/ftpdata.html\).](http://www.esrl.noaa.gov/gmd/dv/ftpdata.html))

was obtained in summer. This inter-annual variations in respiration and photosynthesis fluxes might be attributable to recent local climate variability: 2005 was one of the rainiest years on record in Southern California, encouraging high photosynthetic rates in this usually arid environment, whereas 2004 was significantly drier.

To reconstruct the observed  $\delta^{18}$ O trend we use  $\delta^{18}O_{ff} = 30\%$  ([Affek and Eiler, 2006\)](#page-10-0) but have no independent estimate of the local biospheric  $\delta^{18}$ O end members. We therefore use the flux estimate derived from  $\delta^{13}$ C and  $\Delta^{14}$ C and vary the biospheric end members to yield a best fit of the model results to the  $\delta^{18}O$  observations ([Fig. 1d](#page-2-0); the values used for the end-members are given in Table 4). This results [\(Fig. 1d](#page-2-0)) in a satisfactory reconstruction of the observed trend in variations in 2005. In 2004 we were able to model the average  $\delta^{18}$ O value but not the details of the variations. The large amount of rainfall in Southern California during 2005 is likely to result in small variations in the  $^{18}$ O end-members, thus enabling us to reconstruct the observed variations. In 2004, on the other hand, potential variations in the mixture of water sources used for irrigation probably resulted in complex variations of  $\delta^{18}O$  values of the end-member values. These

variations are not known and therefore not included in our analyses, preventing good reconstruction of the 2004 seasonal pattern.

Finally, we use these flux estimates with the  $\Delta_{47}$  endmembers (given above) to reconstruct the seasonal variations observed in  $\Delta_{47}$  ([Fig. 1](#page-2-0)b). The resulting modelled  $\Delta_{47}$  seasonal curve is therefore independent of the observed  $\Delta_{47}$  data and depends only on the observed concentrations,  $\delta^{13}$ C and  $\Lambda^{14}$ C values and provides a test for the end-member values obtained. The flux rates used yielded a good reconstruction of the seasonal  $\Delta_{47}$  amplitude in 2004 due to relatively large seasonal variations in flux rates. In 2005 flux rates varied much less and resulted in smaller amplitude than observed. A sensitivity test, performed by applying small changes in the  $\Delta_{47}$ end-members, was unable to yield the observed seasonal amplitude. A good fit to the seasonal pattern of 2005 was obtained only by allowing variations in the  $\Delta_{47}$  values of the background air entering the footprint included in our measurements, with slightly lower (0.80 to 0.88 instead of 0.92; Table 4) values in the summer and autumn of 2005. Such variations may be the result of changes in  $\Delta_{47}$  values associated with ocean gas exchange, which are not yet characterized.

Pasadena air is typical of air in urban areas and its isotopic values therefore reflect a complex combination of both anthropogenic and natural local  $CO<sub>2</sub>$  fluxes, on top of the global variations. In spite of these complications, our results indicate that  $\Delta_{47}$  values do not reflect only the expected effect of anthropogenic combustion sources but are affected also by local or regional natural gross fluxes, with a competition between ecosystem respiration and  $CO<sub>2</sub>$  uptake and retro-flux in photosynthesis. Furthermore, the local  $\Delta_{47}$  values reflect also variation in background air  $\Delta_{47}$  values, due to either a change in global background values or to a regional shift in the source of the background air parcels.

This work identifies the existence of measurable temporal, seasonal and diurnal, variations in  $\Delta_{47}$  in atmospheric  $CO<sub>2</sub>$ , which is a prerequisite for it being used as a new isotopic tracer. Although further work must be performed to better characterize the  $\Delta_{47}$  values associated with all the relevant end-members, the preliminary estimates presented here of low  $\Delta_{47}$  values in respiration, very low  $\Delta_{47}$  values in combustion ([Affek and Eiler, 2006\)](#page-10-0) and high values in photosynthetic retro-diffusion, suggest that  $\Delta_{47}$  has the potential to be useful in partitioning photosynthesis, respiration, and fossil fuel combustion, all inseparable using the more common isotopic tracers, such as  $\delta^{13}C$ .

# ACKNOWLEDGMENTS

We thank Eric Klinkner of the Pasadena Water and Power for local electricity production estimates, Michael Bender for the Cape Grim and Barrow air samples and constructive comments as well as Sally Newman and Lisa Welp for helpful discussion. We thank M. Bar-Mathews and two anonymous reviewers for helpful comments. This work is supported by Grants NSF-EAR-0345905, the GPS Davidow Endowment Fund and the David and Lucile Packard Fund.

## **REFERENCES**

- <span id="page-10-0"></span>Affek H. P. and Eiler J. M. (2006) Abundance of mass  $47 \text{ CO}_2$  in urban air, car exhaust, and human breath. Geochimica et Cosmochimica Acta 70, 1–12.
- Bowling D. R., Tans P. P. and Monson R. K. (2001) Partitioning net ecosystem carbon exchange with isotopic fluxes of  $CO<sub>2</sub>$ . Global Change Biology 7(2), 127–145.
- Ciais P., Denning A. S., Tans P. P., Berry J. A., Randall D. A., Collatz G. J., Sellers P. J., White J. W. C., Trolier M., Meijer H. A. J., Francey R. J., Monfray P. and Heimann M. (1997) A threedimensional synthesis study of  $\delta^{18}O$  in atmospheric CO<sub>2</sub>. 1. Surface fluxes. Journal of Geophysical Research-Atmospheres 102(D5), 5857–5872.
- Ciais P., Tans P. P., Trolier M., White J. W. C. and Francey R. J. (1995a) A large northern-hemisphere terrestrial  $CO<sub>2</sub>$  sink indicated by the  ${}^{13}C/{}^{12}C$  ratio of atmospheric CO<sub>2</sub>. Science 269(5227), 1098–1102.
- Ciais P., Tans P. P., White J. W. C., Trolier M., Francey R. J., Berry J. A., Randall D. R., Sellers P. J., Collatz J. G. and Schimel D. S. (1995b) Partitioning of ocean and land uptake of  $CO<sub>2</sub>$  as inferred by  $\delta^{18}$ O measurements from the NOAA Climate Monitoring and Diagnostics Laboratory Global Air Sampling Network. Journal of Geophysical Research-Atmospheres 100(D3), 5051–5070.
- Cuntz M., Ciais P., Hoffmann G., Allison C. E., Francey R. J., Knorr W., Tans P. P., White J. W. C. and Levin I. (2003) A comprehensive global three-dimensional model of  $\delta^{18}O$  in atmospheric CO2: 2. Mapping the atmospheric signal. Journal of Geophysical Research-Atmospheres 108(D17), art. no.-4528.
- Denning A. S., Randall D. A., Collatz G. J. and Sellers P. J. (1996) Simulations of terrestrial carbon metabolism and atmospheric  $CO<sub>2</sub>$  in a general circulation model. 2. Simulated  $CO<sub>2</sub>$  concentrations. Tellus Series B-Chemical And Physical Meteorology 48(4), 543–567.
- Eiler J. M. and Schauble E. (2004)  ${}^{18}O^{13}C^{16}O$  in Earth's atmosphere. Geochimica et Cosmochimica Acta 68(23), 4767–4777.
- Eiler J. M., Schauble E. and Kitchen N. (2003)  ${}^{13}C^{18}O^{16}O$  in air. Geochimica et Cosmochimica Acta 67(18), A86, Suppl. 1.
- Francey R. J. and Tans P. P. (1987) Latitudinal variation in O-18 of atmospheric CO<sub>2</sub>. Nature 327(6122), 495-497.
- Francey R. J., Tans P. P., Allison C. E., Enting I. G., White J. W. C. and Trolier M. (1995) Changes in oceanic and terrestrial carbon uptake since 1982. Nature 373(6512), 326–330.
- Gillon J. S. and Yakir D. (2000) Internal Conductance to  $CO<sub>2</sub>$ Diffusion and  $C^{18}OO$  Discrimination in  $C_3$  Leaves. Plant Physiology 123(1), 201–213.
- Grimmond C. S. B., King T. S., Cropley F. D., Nowak D. J. and Souch C. (2002) Local-scale of carbon dioxide in urban environments: methodological challenges and results from Chicago. Environmental Pollution 116, S243–S254.
- Keeling C. D. (1958) The concentration and isotopic abundances of atmospheric carbon dioxide in rural areas. Geochimica et Cosmochimica Acta 13(4), 322–334.
- Levin I., Kromer B., Schmidt M. and Sartorius H. (2003) A novel approach for independent budgeting of fossil fuel  $CO<sub>2</sub>$  over Europe by  ${}^{14}CO_2$  observations. Geophysical Research Letters 30(23), 2194.
- Lu R. and Turco R. P. (1995) Air pollutant transport in a coastal environment-II. Three-dimensional simulation over Los Angele basin. Atmospheric Environment 29(13), 1499–1518.
- Moriwaki R., Kanda M. and Nitta H. (2006) Carbon dioxide build-up within a suburban canopy layer in winter night. Atmospheric Environment 40, 1394–1407.
- Neiburger M. (1969) The role of meteorology in the study and control of air pollution. Bulletin of the American Meteorological Society 50, 957–965.
- Nemitz E., Hargreaves K. J., Mcdonald A. G., Dorsey J. R. and Fowler D. (2002) Micrometeorological measurements of the urban heat budget and  $CO<sub>2</sub>$  emissions on a city scale. Environmental Science and Technology 36, 3139–3146.
- Newman S., Epstein S., Xu X. and Stolper E. (2004) Changes in concentration and isotopic rations of  $CO<sub>2</sub>$  in air in the Los Angeles basin, California, between 1972 and 2003. Eos Trans. AGU 85(47), Fall Meet. Suppl., Abstract B33A-0235.
- Pataki D. E., Tyler B. J., Peterson R. E., Nair A. P., Steenburgh W. J. and Pardyjak E. R. (2005) Can carbon dioxide be used as a tracer of urban atmospheric transport? Journal Of Geophysical Research-Atmospheres 110(D15), D15102.
- Peylin P., Ciais P., Denning A. S., Tans P. P., Berry J. A. and White J. W. C. (1999) A 3-dimensional study of  $\delta^{18}O$  in atmospheric CO2: contribution of different land ecosystems. Tellus Series B-Chemical and Physical Meteorology 51(3), 642– 667.
- Reid K. H. and Steyn D. G. (1997) Diurnal variations of boundary-layer carbon dioxide in a coastal city—observations and comparison with model results. Atmospheric Environment 31(18), 3101–3114.
- Schauble E. A., Ghosh P. and Eiler J. M. (2006) Preferential formation of  ${}^{13}C-{}^{18}O$  bonds in carbonate minerals, estimated using first-principles lattice dynamics. Geochimica et Cosmochimica Acta 70, 2510–2529.
- Tans P. P., Berry J. A. and Keeling R. F. (1993) Oceanic  ${}^{13}C/{}^{12}C$ observations: a new window on ocean  $CO<sub>2</sub>$  uptake. Global Biogeochemical Cycles 7, 353–368.
- Turnbull J. C., Miller J. B., Lehman S. J., Tans P. P., Sparks R. J. and Southon J. (2006) Comparison of  ${}^{14}CO_2$ , CO, and SF<sub>6</sub> as tracers for recently added fossil fuel  $CO<sub>2</sub>$  in the atmosphere and implications for biological  $CO<sub>2</sub>$  exchange. Geophysical Research Letters 33(1), L01817.
- Wang Z., Schauble E. A. and Eiler J. M. (2004) Equilibrium thermodynamics of multiply-substituted isotopologues of molecular gases. Geochimica et Cosmochimica Acta 68(23), 4779–4797.
- Xu X., Trumbore S. E., Zheng S., Southon J. R., McDuffee K. E., Luttgen M. and Liu J. C. (2007) Modifying a sealed tube zinc reduction method for preparation of AMS graphite targets: Reducing background and attaining high precision. Nuclear Instruments & Methods in Physics Research, Section B 259, 320– 329.
- Yakir D. and Wang X. F. (1996) Fluxes of  $CO<sub>2</sub>$  and water between terrestrial vegetation and the atmosphere estimated from isotope measurements. Nature 380(6574), 515–517.
- Zondervan A. and Meijer H. A. J. (1996) Isotopic characterization of  $CO<sub>2</sub>$  sources during regional pollution events using isotopic and radiocarbon analysis. Tellus Series B-Chemical and Physical Meteorology 48(4), 601–612.

Associate editor: Miryam Bar-Matthews