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2 The Tooth of Time: Cesare Emiliani

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5 It's funny how a seemingly minor event in graduate school can take on deeper meaning later 6 on. We had a day to kill after a long drive from Baltimore to Miami through the Deep South. It 7 was 1967 and we had long hair. Carbonate sedimentologist Bob Ginsburg had come to Johns 8 Hopkins two years earlier, bringing Paleozoic limestones back to life in Florida Bay, the Florida 9 Keys and Great Bahama Bank. He had to spend the day bartering with boat operators but he 10 suggested we attend a talk on forams at the Institute of Marine Science (University of Miami). 11 To be honest, I don't remember much about the substance of the talk, except that the shells of 12 one foram species coil to the left where it is cold and to the right where it is warm—just like 13 American politics. But I remember the speaker, a fast-talking, high-voltage Italian whose 14 presentation was so charged with conviction that for years afterwards I gave all Linnean names 15 an emphatic Bolognese accent in his honour. Only decades after the fact did I begin to appreciate 16 the historic circumstances of Emiliani's talk in Miami.

17 Emiliani studied micropaleontology and published on Cretaceous and Pliocene forams in 18 wartime and post-war northern Italy. He won a Fellowship to University of Chicago in 1948 and 19 was recruited by Harold Urey into his research group, which was dedicated to realizing the 20 dream that "such a transient physical quantity" as the temperature of seawater could be faithfully 21 recorded in rocks for a hundred million years or more (1). The group included engineer Charles 22 McKinney (who supervised the construction of a duplicate mass spectrometer to the one 23 designed by Al Nier in Minneapolis), Urey's graduate student John McCrea and Sam Epstein, a 24 postdoctoral chemist with an interest in geology from Winnipeg by way of McGill. Urey, a 25 Nobel laureate in chemistry in 1934 (age 41), was the most important American Earth scientist of 26 the 20th century, remarkable considering that he devoted only a dozen papers and a tiny fraction 27 of his career to Earth science. As Karl Turekian recently remarked, "Geochemists are often 28 accused of acting like God. There are good reasons for this."

29 In 1946, Urey had been asked (by Paul Niggli) after a talk at ETH in Zürich whether it might 30 be possible, given that rainwater was isotopically lighter than seawater, to distinguish marine and 31 continental carbonates by isotopic analysis. Urey did some calculations and discovered that there 32 was a measurable temperature effect to contend with. "I suddenly found myself with an isotopic 33 thermometer in my hands." he said (2). Not quite. The resolution of their mass spectrometer had 34 to be improved by a factor of ten. And they had to figure out how to extract CO₂ gas from 35 carbonate shells without contamination from the embedded organic matter, a problem Epstein 36 solved by a method still used today (3). By 1951, Urey's group had perfected the measurement 37 of paleotemperatures in ancient carbonates and were able to show that a 150-million-year old 38 belemnite from the Isle of Skye had lived for four winters and three summers at temperatures 39 ranging from 15 to 21°C, and that the winters grew progressively colder during its lifetime. Urey 40 hoped to test the idea that climate change was responsible for the extinction of the dinosaurs, but 41 the results were inconclusive. Emiliani would take Urey's paleothermometer in a new direction.

42 Emiliani reasoned that younger carbonate would be better preserved and that the Pleistocene 43 was a logical target because of large temperature changes associated with the ice ages. He had 44 studied a Pleistocene foraminiferal section near Bologna and had proposed a paleoenvironmental 45 proxy based on foram dimensional variability (prosperity brings uniformity). In 1951, he sampled a Lower Pleistocene section in the Palos Verdes Hills of Los Angeles, finding strong 46 47 oxygen isotope variations. But correlation with existing ice age chronology, hopelessly tied to 48 50-year-old work on the north slope of the eastern Alps-Günz, Mindel, Riss, Würm-was 49 impossible. The same year, he heard a lecture by Hans Pettersson describing recent Swedish 50 innovations in deep-sea piston coring. Cores from the Swedish Deep-Sea Expedition of 1947-48 51 showed quasi-periodic variations in calcium carbonate content, which Gustaf Arrhenius had 52 interpreted in terms of glacial (higher content) and interglacial (lower content) cycles. Arrhenius 53 numbered the glacial (even numbers) and interglacial stages (odd numbers) counting backward 54 from the present interglacial, a scheme later transferred to isotope records. Emiliani saw his 55 chance and visited Göteborg, obtaining samples from a number of Atlantic and Pacific deep-sea 56 cores. These were supplemented by samples Urey obtained from Maurice Ewing at Lamont from 57 the Atlantic and Caribbean. In two of the Swedish cores, much sediment was missing and the 58 core tops, the first samples to be analyzed, turned out to be Miocene and Oligocene in age, based 59 on their benthic forams. Isotopic analysis of these same forams showed that deep-ocean bottom water temperature was 7°C in the Miocene and 10°C in the Oligocene, compared with 2°C today.
Emiliani had found a first-order climate change in deep time in spite of himself (4).

62 Not everything would come so easily. The problem for Pleistocene paleotemperature is that 63 the isotopic composition of seawater is neither uniform nor invariant over time. Epstein and 64 Toshiko Mayeda had shown that the subtropical surface waters are heavy, due to net evaporation, 65 while equatorial waters, estuaries and the Arctic Ocean (a large estuary) are light. During an ice 66 age, the entire ocean becomes heavier in proportion to the volume and degree of isotopic 67 depletion of the ice sheets. Emiliani was acutely aware that temperature and ice volume have 68 complementary isotopic consequences for foraminiferal calcium carbonate, but in what 69 proportions? He went to great lengths to show that the saw-tooth changes in the oxygen isotopes 70 of specific pelagic forams he observed down-hole correlate with changes in foraminiferal 71 speciation and size, then the more established proxy for environmental stress. He studied the 72 depths at which different foraminiferal species live as a means of disentangling temperature from 73 ice volume isotopically. He reasoned that temperature change during glacial-interglacial cycles 74 was greater in surface waters than at depth. Deeper-dwelling species should therefore show less 75 isotopic change than shallower ones if temperature was the main determinant, but all depths 76 should change in lock-step if ice volume was in sole control. He selected four species of 77 planktonic forams that live at different depths and analyzed them independently. Comparing the 78 down-hole curves, Emiliani concluded that about 60% of the variance was temperature and 40% 79 was ice volume (5).

80 This was not the first time invertebrate paleontology played a pivotal role in the struggle to 81 understand the ice ages. In 1836, the celebrated Scottish yachtsman, conchologist and Biblical 82 scholar James Smith (1782-1867) of Glasgow-not to be confused with the English map-83 maker—first used the term "till" in a geological sense, for a "stiff unstratified clay, confusedly 84 mixed with boulders" (6). Together with overlying stratified clays and gravels, till forms what 85 was then called the Diluvium, or Drift, of Newer Pliocene age. (The name Pleistocene was 86 introduced by Lyell in 1839 but he disavowed it. The name was not used during the great 87 controversy over the glacial theory of 1837-1865.) Smith sailed extensively and dredged for 88 shells in the Clyde Estuary and the Arctic North Atlantic. He observed that indigenous (not reworked) marine fauna, mainly molluscan, occur sparingly in the tills and abundantly in the 89 90 stratified Drift, up to hundreds of feet above contemporary sea level. He also pointed out that the

91 Drift fauna of Scotland more closely resembled his Arctic collections than those of the Clyde (7), 92 an inference subsequently confirmed by Edward Forbes (8). The first observation created a major 93 stumbling block for the glacial theory, which predicted a large sea-level fall just when the fauna 94 in the lowland Drift indicated submergence (9). There were no marine fossils in the Alpine 95 glacial deposits and hence the Swiss proponents the glacial theory had no explanation for 96 submergence, a problem not resolved until long after the glacial theory was finally accepted (10). 97 Smith's second observation created a problem for climate physics. It was then widely assumed 98 that global climate could only get colder over time as the Sun grew dimmer through the loss of 99 radiant heat. Europe was experiencing the last advance of the Little Ice Age, so there was 100 historical as well as geological (paleobotanical) evidence for climatic cooling. The faunal 101 evidence for a geologically recent warming implied that climate change was bidirectional, a 102 challenge to physics that led explicitly to the experimental demonstration of the so-called 103 greenhouse effect by the Irish-English physicist John Tyndall in 1861 (11). Never underestimate 104 the power of humble mollusca.

105 Fast forward 100 years to 1961. Teddy Bullard, head of Geodesy and Geophysics at 106 Cambridge University was now convinced that Emiliani's application of Urey's 107 paleothermometer held great promise, and a new stable isotope lab dedicated to 108 paleotemperatures was established in the Quaternary Research unit at Cambridge. The lab would 109 reflect nearly 15 years of technical improvements since Urey's mass spectrometer was installed 110 at Chicago. Setting up the new lab would be the PhD project(!) for a recent Cambridge physics 111 graduate and son of the East Africa field geologist Robert Shackleton. Nick Shackleton was up to 112 the task and by 1967, the year of his PhD (and my trip to Miami), the Cambridge lab was 113 producing data more precise than the Chicago lab (12). But the crux of the problem wasn't 114 precision, it was the same uncertainty plaguing Emiliani, now at the University of Miami: the 115 effect of seawater temperature versus global ice volume on oxygen isotopes in carbonate. Both 116 Emiliani and Shackleton did mass-balance calculations, based on estimated volumes and isotopic 117 compositions for peak glacial and interglacial ice sheets: Emiliani's estimate of the mean 118 isotopic depletion of the Laurentide ice sheet was -15% PDB, less than half the value calculated 119 by Eric Olausson in Göteborg (based on Wili Dansgaard's data from Greenland), subsequently 120 confirmed by Shackleton (13). The result was that Emiliani's calculation underestimated the ice-121 volume effect by a factor of two.

122 Shackleton reasoned that abyssal bottom water could not have been much colder during ice 123 ages because it is close to the freezing point today. So he measured the compositions of 124 coexisting planktonic and benthic forams in two samples spanning the last glacial-interglacial 125 transition in a tropical South Pacific deep-sea core. He supplemented his data with paired 126 benthic-planktonic results from Atlantic and Caribbean cores published by Emiliani. Because 127 benthic forams are larger and less abundant, fewer individuals contributed to each analysis: the 128 result was scatter in the data because bioturbation can disturb the mean age of a few shells more 129 readily than a few hundred, leading to asynchroneity between the two populations in a sample. 130 Nevertheless, the best-fit line in a benthic-planktonic cross-plot of the data had a slope close to 131 unity—as much isotopic change in bottom water as surface water (14). This meant that ice 132 volume was the dominant factor.

133 Shackleton's paper (14) appeared in Nature on 01 July 1967 and was a direct attack on 134 Emiliani's assertion that foram isotope change was predominantly a paleotemperature record. 135 Despite the uncertainty as to whether I heard Emiliani on my first trip to Miami with Ginsburg 136 (Spring 1967) or my second (Spring 1968), it is likely that Emiliani knew what was coming, 137 even if it hadn't yet arrived. Moreover, the dagger had a twist. Shackleton needed to explain why 138 Emiliani's planktonic species from different depths did not change in lock-step, as they should 139 do if the composition of seawater, not its temperature, was in control. Recalling an argument put 140 forward by oceanographer Wally Broecker, Shackleton suggested that because the glacial ocean 141 was saltier, planktonic forams had risen into warmer waters due to buoyancy. Emiliani appears to 142 have neglected this factor and it must have galled the foram specialist to be caught out by a 143 newly-minted PhD geochemist. But then Shackleton showed his class. In conclusion, he wrote, 144 Emiliani's curves become even more valuable because they constitute a direct record of the ice ages, not an indirect record through seawater temperature (14). 145

The ice-volume only paradigm prevailed for twenty years before the tide turned again in the 1990s as data from terrestrial climate records (palynology, tropical snow-line elevations, noble gases in tropical groundwater), porewaters from deep-sea cores, and new paleothermometers (Mg/Ca, Sr/Ca, alkenone unsaturation index) confirmed that the tropics did cool significantly during glacial maxima after all. Today the respective contributions of temperature and ice volume to the foram isotope record are thought to lie about halfway between Shackleton's estimate and Emiliani's. 153 Both went on to greatness. In 1967, Emiliani was appointed Chairman of Geology and 154 Geophysics at the Marine Institute (later Rosenstiel School of Marine and Atmospheric Sciences) 155 and of Geological Sciences on the main campus of the University of Miami, positions he held 156 until his retirement in 1993. He was instrumental in the establishment of the Deep-Sea Drilling 157 Project by NSF in 1968, with its emphasis on paleoenvironmental records (15). Versed in the 158 classics and virtually all of science and its history, he was a spell-binding teacher and his 159 textbook, Planet Earth (1992), and The Scientific Companion (1987, 1995) filled a niche now 160 occupied by Wikipedia.

161 Shackleton became a key participant in CLIMAP, a multi-institutional NSF project to 162 reconstruct LGM (last glacial maximum) sea-surface temperatures globally, as a boundary 163 condition for climate models. Resulting collaborations at Lamont led to the recognition with 164 magnetostratigrapher Neil Opdyke of Pacific Core V28-238 from the Solomon Plateau, which 165 yielded a beautiful isotope record of 22 stages, tied for the first time to the Brunhes-Matuyama 166 geomagnetic reversal (0.78 Ma) in stage 19 (16). The isotope record from V28-238 could be 167 correlated stage by stage with Emiliani's records from the opposite side of the globe (17). 168 CLIMAP also spawned Shackleton's collaboration with Jim Hays (Lamont) and John Imbrie 169 (Brown), resulting in the spectral analysis of deep-sea records that resurrected the Milankovitch 170 theory as the pacemaker of the ice ages (18). This turn of events brought joy to Emiliani, who 171 had been an outspoken supporter of the Milankovitch theory during its eclipse (19).

Neither Emiliani nor Shackleton are alive to correct my reconstruction of events, but Bob Ginsburg remembers. His off-hand suggestion that a bunch of travel-weary graduate students go to a talk on forams was anything but. He knew Emiliani from his own graduate-school days at Chicago. He knew what we were in for, if we took the bait. Thanks Bob.

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177 References and notes:

- 178 (1) Urey, H.C., 1948, Oxygen isotopes in nature and in the laboratory: Science, v. 108, p. 489-496.
- 179 (2) Emiliani, C., 1995, Two revolutions in the Earth Sciences: Terra Nova, v. 7, p. 587-597.
- 180 (3) Epstein, S., 1997, The role of stable isotopes in geochemistries of all kinds: Annual Reviews of Earth
- and Planetary Sciences, v. 25, p. 1-21.

- (4) Emiliani, C., 1954, Temperatures of Pacific bottom waters and polar superficial waters during the
 Tertiary. Science, v. 119, p. 853-855.
- 184 (5) Emiliani, C., 1955, Pleistocene temperatures: Journal of Geology, v. 63, p. 538-578; Emiliani, C.,
 185 1995, op. cit.
- (6) Smith, J., 1836, On indications of changes in the relative level of Sea and Land in the West of
 Scotland: Proceedings of the Geological Society, London, v. 2, p. 427-429.
- (7) Smith, J., 1837, On the Climate of the newer pliocene tertiary period: Proceedings of the Geological
 Society, London, v. 3, 118-119. Smith's collected papers on geology were conveniently republished
 as Researches in Newer Pliocene and Post-Tertiary Geology. John Gray, Glasgow, 199 p., 1862.
- 191 (8) Forbes, E., 1846, On the connexion between the distribution of the existing fauna and flora of the
- British Isles and the geological changes which have affected their area, especially during the epoch of
 the Northern Drift: Memoirs of the Geological Survey of Great Britain, v. 1, p. 336-403.
- (9) Maclaren, C., 1842, The glacial theory of Prof. Agassiz: American Journal of Science and Arts, v. 42,
 p. 346-365.
- 196 (10) Shaler, N.S., 1874, Preliminary report on the Recent changes of level on the coast of Maine:
- 197 Memoirs of the Boston Society of Natural History, v. 2, p. 321-323, 335-340; Jamieson, T.F., 1882,
- 198 On the cause of the depression and re-elevation of the land during the glacial period: Geological
- 199 Magazine, v. 9, p. 400-407, 457- 466.
- (11) Tyndall, J., 1861, On the absorbtion and radiation of heat by gases and vapours, and on the physical
 connexion of radiation, absorbtion, and conduction the Bakerian Lecture: Philosophical Magazine,
 Ser. 4, v. 22, p. 169-194, 273-285; Tyndall, J., 1863, On radiation through the Earth's atmosphere:
 Philosophical Magazine, Ser. 4, v. 25, p. 200-2006. Tyndall's collected papers on atmospheric
- science were conveniently republished as Contributions to Molecular Physics in the Domain of
 Radiant Heat. Longman's, Green and Co., London, 446 p., 1872.
- 206 (12) Elderfield, H., 2006, Professor Sir Nicholas Shackleton: The Independent, Obituaries, 08 February.
- 207 (13) Emiliani, C., 1955, op. cit.; Emiliani, C., 1966, Isotopic paleotemperatures: Science, v. 154, p. 851-
- 208 857; Olausson, E., 1963, Evidence of climatic changes in North Atlantic deep-sea cores, with remarks
- on isotopic paleotemperature analysis: Progress in Oceanography, v. 3, p. 221-252; Dansgaard, W.,
- 210 1961, The isotopic composition of natural waters, with special reference to the Greenland ice cap:
- 211 Meddelelser om Grønland, v. 165(2), 120 p.; Shackleton, N., 1967, Oxygen isotope analyses and
- 212 Pleistocene temperatures re-assessed: Nature, v. 215, p. 15-17; Dansgaard, W., and Tauber, H., 1969,
- Glacier oxygen-18 content and Pleistocene ocean temperatures: Science, v. 166, p. 499-502. For a
- recent estimate, supportive of Olausson and Shackleton, see Sima, A., Paul, A., Schulz, M., and

- 215 Oerlemans, J., 2006, Modeling the oxygen isotopic composition of the North American Ice Sheet and
- 216 its effect on the isotopic composition of the ocean during the last glacial cycle. Geophysical Research
- 217 Letters, v. 33, L15706, doi: 10.1029/2006GL026923
- (14) Shackleton, N., 1967, Oxygen isotope analyses and Pleistocene temperatures re-assessed: Nature, v.
 219 215, p. 15-17.
- (15) Shor, E.N., 1985. A chronology from Mohole to JOIDES. in Drake, E.T., and Jordan, W.M., eds.,
 Geologists and ideas: a history of North American geology: Geological Society of America
 Centennial Special Volume 1, p. 391-399.
- (16) Shackleton, N.J., and Opdyke, N.D., 1973, Oxygen isotope and paleomagnetic stratigraphy of
 equatorial Pacific Core V28-238: oxygen isotope temperatures and ice volumes on a 10⁵ year and 10⁶
 year scale: Quaternary Research, v. 3, p. 39-55.
- (17) Emiliani, C., and Shackleton, N.J., 1974, The Brunhes epoch: isotopic paleotemperatures and
 geochronology: Science, v. 183, p. 511-514.
- (18) Hays, J.D., Imbrie, J., and Shackleton, N.J., 1976, Variations in the Earth's orbit: pacemaker of the
 ice ages: Science, v. 194, p. 1121-1132.
- 230 (19) Emiliani, C., and Geiss, J., 1959, On glaciations and their causes: Geologische Rundschau, v. 46, p.
- 576-601; Emiliani, C., 1969, Interglacial high sea levels and the control of Greenland ice by the
- precession of the equinoxes: Science, v. 166, p. 1503-1504; Emiliani, C., 1978, The cause of the ice
- ages: Earth and Planetary Science Letters, v. 37, p. 349-352.